

1D constrained surface reactions on anisotropic Au(110)

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Au(110) surfaces with missing-row reconstruction exhibit promising effect for on-surface chemical processes, due to the one dimensional (1D) geometric confinement. Linear alkane polymerization have been achieved on Au(110) surfaces through the terminal methyl C-H activation [1]. Compared with Pt(110), which has similar missing-row reconstruction, the Au(110) surface exhibits relatively weak interactions with dehydrogenated hydrocarbon species, suppressing C-C bond breaking and fragmentation [2]. Besides linear alkanes, the concept of “1D constrained surface reactions” can be extended to other molecular systems. C-H bond activation of oligophenylenes resulting in the preferential formation of linear polyphenyl wires over branched analogous has been demonstrated on Au(110) surfaces. Owing to the 1D spatial constraint, the equivalent C-H bonds of *p*-6P molecules are differentiated with various reaction probabilities. Consequently, linear polymerization of *p*-6P has been achieved via C-C bonding at the meta-sites, with the kink formation every 6 phenylene units (Fig. 1) [3]. Furthermore, the “1D constrained surface reactions” have been successfully applied for the decarboxylation of linear fatty acids, resulting in the polymerization at their terminal ends and the formation of oxygen-free aliphatic hydrocarbons at mild temperatures. The 1D constrained surface reactions beyond Au(110) surfaces will also be discussed in this talk [4].

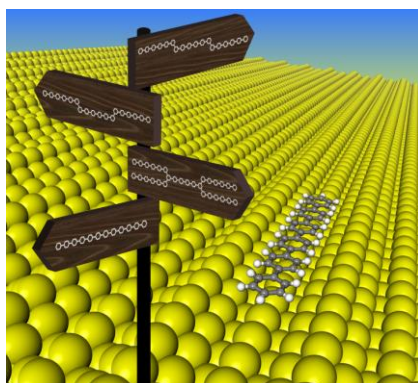


Fig. 1 Controllable fabrication of linear polyphenyl wires by direct C-H activation and C-C coupling of para-sexiphenyl (*p*-6P) has been achieved on an anisotropic Au(110) surface. Owing to the one-dimensional spatial constraint at the surface, the equivalent C-H bonds of *p*-6P molecules are differentiated with various reaction probabilities.

References:

- [1] D. Y. Zhong et al., *Science* **334**, 213 (2011).
- [2] Z. Y. Cai et al., *ChemPhysChem* **16**, 1356 (2015).
- [3] Z. Y. Cai et al., *J. Phys. Chem. C* **120**, 6619 (2016).
- [4] M. Z. Liu et al., *Nat. Commun.* **8**, 14924 (2017).